visocolor®ECO



Chlorine 1

Test kit for performing colorimetric tests on free and total chlorine in drinking water, swimming pools, and water reservoirs

Method:

At a pH value of 6.2 to 6.5 in a phosphate buffered system, free chlorine reacts with N,N-diethyl-1,4-phenylene diamine (DPD) and forms a red-violet dye. In the presence of iodide ions, the content of total chlorine (free and combined chlorine together) can be determined.

Measurement range:

0.1-2.0 mg/L Cl₂

Contents of test kit (*refill pack):

sufficient for 150 tests

6 g Cl₂-1* 30 mL Cl₂-2*

- measuring spoon 70 mm*
- 2 screw-plug measuring glasses1 slide comparator
- color chart
- 1 plastic syringe 5 mL 1 instructions for use*

Hazard warning:

Information regarding safety can be found on the box' label and in the safety data sheet. You can download the SDS from www.mn-net.com/SDS.

Instructions for use:

also refer to the pictogram on the back of the color chart

Free chlorine a)

Pour a 5 mL water sample into each of the measuring glasses using the plastic syringe (the pH value of the sample must be between pH 4 and 8.) Place a measuring glass on position A in the comparator. and

Only add the reagent to measuring glass B.

- 2. Add 1 level measuring spoon of Cl2-1, seal the glass and shake well for 20 s.
- 3. Open the glass after 1 min and place it on position B in the comparator.
- Slide the comparator until the colors match in the inspection hole on top. 4. the measurement reading in the recess on the comparator reed. Mid-values can be estimated.
- b) **Total chlorine**
- Add 3 drops of Cl₂-2. Seal the glass and mix. 6
- Open the glass after **2 min**, place it on position B in the comparator and read off the chlorine value as described above.
- 7 After use, rinse out both measuring glasses thoroughly and seal them.

c) **Combined chlorine**

The content of combined chlorine can be calculated as difference of total and free chlorine.

The reagents can be used also for the $photometric\ evaluation$ with photometers PF-3/PF-12 and PF-12 $^{\text{Plus}}.$ This technique can be used also for analyzing sea water.

Disposing of the samples:

Information regarding disposal can be found in the safety data sheet. You can download the SDS from www.mn-net.com/SDS. Interferences:

The determination of free chlorine measures bromine, bromamine, chloramine, iodine and, in part, chlorine dioxide as well. Higher manganese compounds simulate free chlorine. Chlorine concentrations above 10 mg/L can bleach the red reaction color

(low results).

Rinse glass tubes several times thoroughly. Residues of ${\rm Cl_2-2}$ can cause higher values for free chlorine.

Note:

Determination of bromine besides chlorine: If chlorine is present in the sample, it can be destroyed by adding a spatula of glycine (approx. 20 mg) to 25 mL sample. The sample for the bromine determination is taken from this 25 mL sample. The sample for the bromine d solution. Result in mg/L Cl_2 x 2.25 = mg/L Br_2 .

Conversion table:

Conversion table.					
mg/L Cl ₂	mg/L CIO ₂	mg/L OCI	mg/L NaOCI	mg/L Br ₂	$mg/L I_2$
0.1	0.2	0.1	0.2	0.2	0.4
0.2	0.4	0.3	0.4	0.5	0.7
0.3	0.6	0.4	0.6	0.7	1.1
0.4	0.8	0.6	0.8	0.9	1.4
0.6	1.1	0.9	1.3	1.4	2.1
0.9	1.7	1.3	1.9	2.0	3.2
1.2	2.3	1.7	2.5	2.7	4.3
2.0	3.8	2.9	4.2	4.5	7.2

For swimming pools (in Germany) please note:

If the content of free chlorine is below 0.3 mg/L, add some chlorinating reagent. If the content is above 0.6 mg/L, add fresh water. The ideal pH value is 7.4.

Storage:

Store the test kit in a cool (< 25 °C) and dry place.